

Answer on Question #67388, Physics / Solid State Physics

Iron atom has magnetic moment 2.2 mB. Calculate its saturation magnetisation?

Find: M_s – ?

Given:

$$p = 2.2 \times 9.27 \times 10^{-24} \text{ J} \times \text{T}^{-1}$$

Solution:

Saturation magnetisation M_s is a magnetic moment p per unit volume V .

We find the magnetic moment per cubic cell (which contains 2 Fe atoms in bcc Fe metal) and divide this moment by the cell volume to obtain the magnetization in units of emu/cm^3 .

$$M_s = \frac{2 \times 2.2 \times 9.27 \times 10^{-24}}{(2.87 \times 10^{-24})^3} = 1725 \text{ emu}/\text{cm}^3$$

Answer:

$$M_s = 1725 \text{ emu}/\text{cm}^3$$

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