

### Answer on Question #64423, Physics / Atomic and Nuclear Physics

What is approximate density of nuclear matter from which all nuclei are made? Also compare with that water.

**Answer:**

The volume of the nucleus approximated as a uniform sphere of radius R.

$$V = \frac{4}{3} \pi R^3 = \frac{4}{3} \pi (R_0^3 A)$$

The density  $\rho_0$  of nuclear matter

$$\rho_0 = A / V = A / \frac{4}{3} \pi (R_0^3 A) = 1 / \frac{4}{3} \pi (1.2 \text{ fm})^3 = 0.14 \text{ nucleons/ fm}^3$$

The mass of a nucleon is  $1.7 \times 10^{-27} \text{ kg}$

The mass density  $\rho_m$  of nucleon matter is then

$$\rho_m = (0.14 \text{ nucleons/ fm}^3) (1.7 \times 10^{-27} \text{ kg}) (1 \text{ fm} / 10^{-15} \text{ m})^3 = 2.4 \times 10^{17} \text{ kg/m}^3$$

$2.4 \times 10^{14}$  times the density of water.