

Answer on Question #64197, Physics / Molecular Physics | Thermodynamics

If the pressure in a steam boiler as shown by the gauge is 3.6 bar, what is the temperature of the steam?

Find: T- ?

Given:

$$T_0=273 \text{ K}$$

$$p_0=1 \text{ bar}$$

$$p=3.6 \text{ bar}$$

Solution:

The steam is in the boiler and the volume of steam is not changed.

Gas law at constant volume:

$$\frac{p}{T} = \frac{p_0}{T_0} \quad (1)$$

$$\text{Of (1)} \Rightarrow T = \frac{p}{p_0} T_0 \quad (2)$$

$$\text{Of (2)} \Rightarrow T=982.8 \text{ K}$$

$$T=t^\circ\text{C}+273 \quad (3)$$

$$\text{Of (3)} \Rightarrow t=709.8^\circ\text{C}$$

Answer:

$$982.8 \text{ K (709.8}^\circ\text{C)}$$

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