

Answer on Question #63664, Physics / Atomic and Nuclear Physics

Can a 1 kg of stone be converted in to energy according to $E=mc^2$?

Answer:

The formula $E=mc^2$ shows that matter and energy are interrelated, matter can be converted into energy and energy can be converted into mass.

$$E=mc^2$$

Where: E is Energy in Joules (J); m is mass in kilograms (kg); c is the speed of light in meters per second (ms^{-1}) which is $3.0 \times 10^8 \text{ ms}^{-1}$.

If a stone weighing 1 kg were completely converted into energy it would release so much energy:

$$E = 1 \text{ kg} \times 9.0 \times 10^{16} \text{ m}^2\text{s}^{-2} = 9.0 \times 10^{16} \text{ J} = 5.63 \times 10^{35} \text{ eV}$$

If the stones could converted directly to energy, we could be faced with inexhaustible amounts of fuel.

However, it is practically impossible to implement.

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