## Question:

The mean speed of oxygen molecule is $450 \mathrm{~m} / \mathrm{s}$. If the radius of an oxygen molecule is $1.8 \times 10^{\wedge}$ _ 10 m , calculate mean time between two successie collisions and mean free path. Take $\mathrm{n}=3 \mathrm{x}$ $10^{\wedge} 25 \mathrm{~m}$.

Answer:
$l=\frac{1}{\sqrt{2} \pi d^{2} n}=\frac{1}{4 \sqrt{2} \pi r^{2} n}=5.789 \cdot 10^{-8} \mathrm{~m} ;$
$t=\frac{l}{v}=1.286 \cdot 10^{-10} \mathrm{sec} ;$

