## Answer on Question \#59740-Physics- Electromagnetism

A typical polar molecule has a dipole moment of the order of magnitude $1.6 \times 10-29 \mathrm{Cm}$. Calculate the energy required to reverse its direction to be opposite to an electric field of intensity $106 \mathrm{~V} / \mathrm{m}$

## Solution

The potential energy of dipole moment is

$$
U=-p E \cos \theta
$$

At the initial state $\theta=0$ :

$$
U_{i}=-p E \cos 0=-p E
$$

At the final state $\theta=\pi$ :

$$
U_{f}=-p E \cos \pi=p E
$$

The requires energy is

$$
W=U_{f}-U_{i}=p E-(-p E)=2 p E=2 \cdot 1.6 \cdot 10^{-29} \cdot 10^{6}=3.2 \cdot 10^{-23} \mathrm{~J}
$$

