

## Answer on Question #51237, Physics, Solid State Physics

The Debye temperature for silver is 225 K. Calculate the highest possible frequency for lattice vibrations in silver and its molar heat capacity at 10 K and 500 K.

### Solution:

The maximum frequency of vibration of the atoms of solid bodies

$$\nu_{\max} = \frac{k_B T_D}{h} = \frac{1.38 \cdot 10^{-23} \text{ J/K} \cdot 225 \text{ K}}{6.62 \cdot 10^{-34} \text{ J} \cdot \text{s}} = 4.7 \cdot 10^{12} \text{ Hz} \quad (1)$$

where  $k_B = 1.38 \cdot 10^{-23} \text{ J/K}$  is the Boltzmann constant;  $h = 6.62 \cdot 10^{-34} \text{ J} \cdot \text{s}$  is the Planck constant;  $T_D$  is the Debye temperature.

The molar heat capacity is given by Eq.(2)

$$C_V(T) = 3Nk_B \left( \frac{3}{x_D^3} \int_0^{x_D} \frac{x^4 e^x}{(e^x - 1)^2} dx \right) \quad (2)$$

where  $3N$  is number of normal modes;  $x_D = T_D / T$ .

Then

$$C_V(10) = 3Nk_B \left( \frac{3}{(225/10)^3} \int_0^{225/10} \frac{x^4 e^x}{(e^x - 1)^2} dx \right) = 0.0205Nk_B$$

$$C_V(500) = 3Nk_B \left( \frac{3}{(225/500)^3} \int_0^{225/500} \frac{x^4 e^x}{(e^x - 1)^2} dx \right) = 2.9698Nk_B$$

**Answer:**  $\nu_{\max} = \frac{k_B T_D}{h} = 4.7 \cdot 10^{12} \text{ Hz}$ ;  $C_V(10) = 0.0205Nk_B$ ;  $C_V(500) = 2.9698Nk_B$