

Question#18832

The temperature of an electronic oven is 167.3 C. The temperature at the outer surface of the oven in the kitchen is 48.8 C. The oven (surface area = 1.7 m²) is insulated with material that has a thickness of 0.02 m and a thermal conductivity of 0.04 J/(s m C). How much energy in Joules will be used to operate the oven over 3 hours?

Solution:

Let:

$$t_0 = 48.8 \text{ }^\circ\text{C}$$

$$t_1 = 167.3 \text{ }^\circ\text{C}$$

$$S = 1.7 \text{ m}^2$$

$$d = 0.02 \text{ m}$$

$$k = 0.04 \text{ J/sm}^\circ\text{C}$$

$$T = 3 \text{ hours} = 10800 \text{ sec}$$

$$Q = ?$$

$$Q = qT, \text{ where } q - \text{ the heat flux, } T - \text{ time}$$

According to the Fourier's law (the law of heat conduction):

$$q = k \frac{S \Delta t}{d}$$

$$\Delta t = t_1 - t_0$$

$$q = k \frac{S(t_1 - t_0)}{d}$$

$$Q = kT \frac{S(t_1 - t_0)}{d}$$

$$Q = 0.04 * 10800 * \frac{1.7(167.3 - 48.8)}{0.02} = 2559600 \text{ J}$$

Answer: 259600 J (256.6 KJ)