

### Answer on Question #11567-Physics-Molecular Physics-Thermodynamics

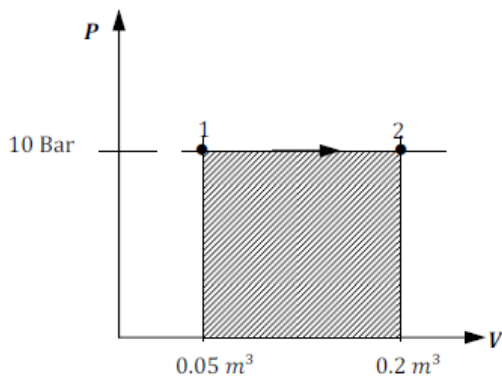
A certain fluid at 10 bar is contained in a cylinder behind a piston, the initial volume being  $0.05 \text{ m}^3$ . Calculate the work done by the fluid when it expands reversibly:

- i. at constant pressure to a final volume of  $0.2 \text{ m}^3$
- ii. according to a linear law to a final volume of  $0.2 \text{ m}^3$  and a final pressure of 2 bar
- iii. according to a law  $pV = \text{constant}$  to a final volume of  $0.1 \text{ m}^3$
- iv. according to a law  $pV^3 = \text{constant}$  to a final volume of  $0.06 \text{ m}^3$
- v. according to a law  $p = (a/v^2) - (b/v)$  to a final volume of  $0.1 \text{ m}^3$  and a final pressure of 1 bar where A and B are constants.

Sketch all processes on a p-V diagram.

#### Solution

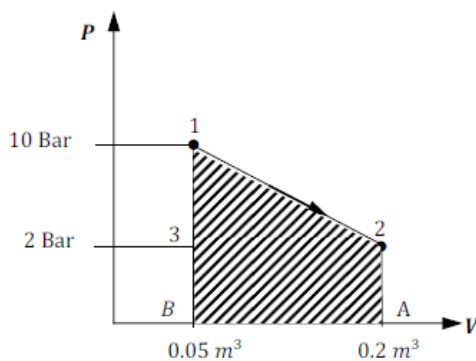
i.



For reversible non-flow process we have

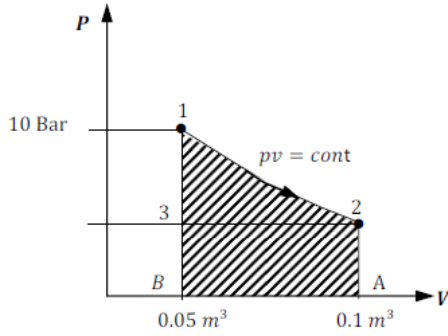
$$W = \text{shaded area} = \int_{V_1}^{V_2} p dV = p(V_2 - V_1) = 1000 \text{ kPa} \cdot (0.2 - 0.05) \text{ m}^3 = 150 \text{ kJ}.$$

ii.



$$W = \text{shaded area} = \left( \frac{p_1 + p_2}{2} \right) |V_2 - V_1| = \left( \frac{10 + 2}{2} \right) 10^5 |0.2 - 0.05| = 90 \text{ kJ}.$$

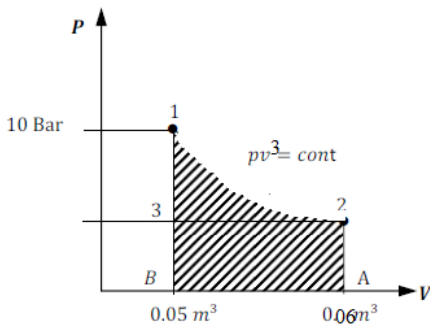
iii.



$$W = \text{shaded area} = \int_1^2 p dV = \int_1^2 \frac{p_1 V_1}{V} dV.$$

$$W = p_1 V_1 \ln \frac{V_2}{V_1} = 1000 \cdot 0.05 \cdot \ln \left( \frac{0.1}{0.05} \right) = 34.7 \text{ kJ}.$$

iv.



$$W = \frac{p_2 V_2 - p_1 V_1}{(1 - 3)}.$$

Where  $p_2 = p_1 \left( \frac{V_1}{V_2} \right)^3 = 10 \left( \frac{0.05}{0.06} \right)^3 = 5.787 \text{ bar}.$

$$W = \frac{5.787 \cdot 0.06 - 10 \cdot 0.05}{-2} 10^5 = 7640 \text{ J}.$$

v.

$$W = \int_1^2 \left( \frac{a}{V^2} - \frac{b}{V} \right) dV = a \left( \frac{1}{V_1} - \frac{1}{V_2} \right) + b \ln \frac{V_1}{V_2}.$$

$$10 \text{ bar} = \left( \frac{a}{(0.05 \text{ m}^3)^2} \right) - \left( \frac{b}{0.05 \text{ m}^3} \right); 1 \text{ bar} = \left( \frac{a}{(0.1 \text{ m}^3)^2} \right) - \left( \frac{b}{0.1 \text{ m}^3} \right).$$

$$a = 0.04 \text{ bar m}^6; b = 0.3 \text{ bar m}^3.$$

Thus,

$$W = 0.04 \text{ bar m}^6 \left( \frac{1}{0.05 \text{ m}^3} - \frac{1}{0.1 \text{ m}^3} \right) + 0.3 \text{ bar m}^3 \ln \frac{0.05 \text{ m}^3}{0.1 \text{ m}^3} = 19.2 \text{ kJ}.$$

<http://www.AssignmentExpert.com/>