## Answer on Question \#81681 - Math - Algebra

## Question

$30 \%$ silver is mixed with 200 g of a $10 \%$ silver alloy. How much of the $30 \%$ alloy must be used to obtain an alloy that is $24 \%$ silver?

## Solution

$x \mathrm{~g}=$ mass of $30 \%$ alloy;
$0.3 \times \mathrm{g}=$ mass of silver in $30 \%$ alloy;
$0.3 x+0.1 \times 200=0.24 \times(200+x)$ is the total mass of silver after mixing;
$0.3 x+20=48+0.24 x$
$0.3 x-0.24 x=48-20$
$0.06 x=28$
$x=\frac{28}{0.06}$
$x=466.7$

## Answer:

466.7 g of the $30 \%$ alloy must be used to obtain an alloy that is $24 \%$ silver.

