

Question: When polluted water begins to flow into an unpolluted pond, the concentration of pollutant, c , in the pond at t minutes is modelled by $c(t) = \frac{27t}{10000+3t}$, where c is measured in kilograms per cubic metre. Determine the rate at which the concentration is changing after

- a) 1 h
- b) one week.

Solution. We need to find the instantaneous rate of change for $c(t)$. First, apply the formula for average rate of change:

$$\frac{c(t+h) - c(t)}{h} = \frac{1}{h} * \left(\frac{27t + 27h}{10000 + 3t + 3h} - \frac{27t}{10000 + 3t} \right)$$

Bring the expression at the right to a common denominator and simplify:

$$\frac{c(t+h) - c(t)}{h} = \frac{1}{h} * \frac{270000h}{(10000 + 3t + 3h)(10000 + 3t)} = \frac{270000}{(10000 + 3t + 3h)(10000 + 3t)}.$$

We now find the instantaneous rate of change as the limit of average rate of change as h tends towards zero:

$$\begin{aligned} \lim_{h \rightarrow 0} \frac{c(t+h) - c(t)}{h} &= \\ &= \lim_{h \rightarrow 0} \frac{270000}{(10000 + 3t + 3h)(10000 + 3t)} = \frac{270000}{(10000 + 3t)(10000 + 3t)} \\ &= \frac{270000}{(10000 + 3t)^2}. \end{aligned}$$

Now let us find the rate of change for values of t required in the question:

- a) 1 h = 60 min. $t = 60$

Rate of change:

$$\frac{270000}{(10000 + 3 * 60)^2} = \frac{270000}{10180^2} \approx 0.0026 \text{ (kg/m}^3\text{/min)}$$

- b) 1 week = $7 * 24$ h = $7 * 24 * 60$ min = 10080 min. $t = 10080$

Rate of change:

$$\frac{270000}{(10000 + 3 * 10080)^2} = \frac{270000}{40240^2} \approx 0.0001667 \text{ (kg/m}^3\text{/min)}$$

Answer. The rate of change is:

- a) After 1 h – approximately 0.0026 kg/m³/min.
- b) After one week – approximately 0.00017 kg/m³/min.