

Question #85186, Chemistry / General Chemistry | for completion

A student obtained the following values for the percentage of Br in a bromide sample: 21, 20, 27, 22, and

19. What should be reported as the best estimate of the Br content of the unknown sample?

Answer:

73-----x

27-----80

X=216.

This metall maybe Pb (lead), because the closest atomic mass to 216 element is Lead.

Answer provided by www.AssignmentExpert.com