Question \#85133, Chemistry / General Chemistry | for completion
if 25 J are required to change the temperature of 5.0 g of a substanceA by 2.0 degrees celsius, what is the specific heat of a substance $A$

Solution:
$\mathrm{m}=5 \mathrm{~g}=0.005 \mathrm{~kg}$
$\mathrm{c}=\frac{\mathrm{Q}}{\mathrm{m} \cdot \Delta \mathrm{T}}=\frac{25 \mathrm{~J}}{0.005 \mathrm{~kg} \cdot 2 K}=250 \frac{\mathrm{~J}}{\mathrm{~kg} \cdot K}$
Answer: $250 \frac{\mathrm{~J}}{\mathrm{~kg} \cdot K}$

