For the following exothermic reaction at equilibrium

 $A(g) + B(g) \rightleftharpoons 2 C(g)$

Which of the following conditions will cause the reaction to produce more products?

Select as many answers as are appropriate however marks will be deducted for incorrect guesses.

Increasing partial pressure of "B" Decreasing the volume Lowering the temperature Increasing partial pressure of "C" Increasing partial pressure of "A" Increasing the volume Raising the temperature

Answer:

Increasing partial pressure of "B" Decreasing the volume Lowering the temperature Increasing partial pressure of "A"

Answer provided by www.AssignmentExpert.com