Calculate mass fraction of K in KNO 3 , if m (weight) of $\mathrm{K}=45$ grams, m (weight) of $\mathrm{N}=23 \mathrm{grams}$ and n (amount of substance) of $\mathrm{O}=0,58 \mathrm{~mol}$

## Solution.

$m(0)=0.58 * 16=9.28 \mathrm{~g}$
$w(K)=45 / 77.28=0.582$ (58.2\%)

## Answer. 58.2\%

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