

Answer on Question #82179 – Chemistry – General Chemistry

Question

According to Le Chatelier's principle, increased pressure will _____.

- A. always shift the equilibrium to a higher reaction rate
- B. shift the equilibrium toward the side with the largest number of gas molecules
- C. destroy the equilibrium if there are no gas molecules
- D. shift the equilibrium away from the side with the largest number of gas molecules
- E. destroy the equilibrium if the number of gas molecules is equal on both sides

Answer: D. shift the equilibrium away from the side with the largest number of gas molecules.

Answer provided by www.AssignmentExpert.com