Answer on the Question \#81179 - Chemistry - General Chemistry
This substance is $\mathrm{KMnO}_{4}$. The reaction has view:
$\mathrm{H}_{2} \mathrm{O}+\mathrm{KMnO}_{4}=4 \mathrm{MnO}_{2}+3 \mathrm{O}_{2}+4 \mathrm{KOH}$
The easiest way to get $\mathrm{O}_{2}$ from water is to using electrolisys, but there wouldn't be any material.

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