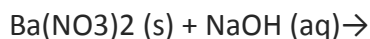
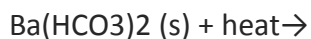
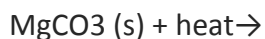
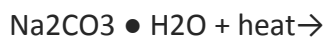
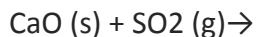
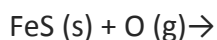
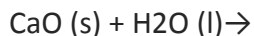
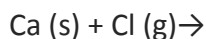


Answer on Question #79090, Chemistry/ General Chemistry

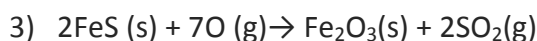
Complete and Balance equation:



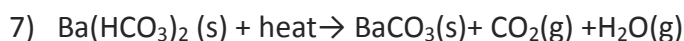
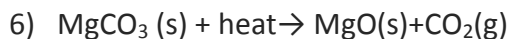
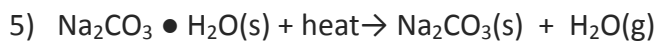
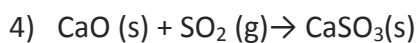
Answer



Usually other notation of chlorine is used: $\text{Ca (s)} + \text{Cl}_2 \text{ (g)} \rightarrow \text{CaCl}_2 \text{ (s)}$



Usually other notation of oxygen is used: $4\text{FeS (s)} + 7\text{O}_2 \text{ (g)} \rightarrow 2\text{Fe}_2\text{O}_3 \text{ (s)} + 4\text{SO}_2 \text{ (g)}$



If more heat is added to $\text{Ba(HCO}_3)_2$ further decomposition takes place:

