

Answer on Question #78397 - Chemistry - Physical Chemistry

Question: A reaction is catalyzed by an enzyme with parameters $k=75 \text{ s}^{-1}$, $K_M=24 \cdot 10^{-6} \text{ M}$ and concentration $[E]_0= 2.5 \cdot 10^{-6} \text{ M}$. The initial concentration of the substrate is $[S]_0= 8 \cdot 10^{-4} \text{ M}$. Estimate initial reaction rate. Find the substrate's concentration providing 4-fold decreasing of the reaction rate.

Solution:

$$v_0 = \frac{k \cdot [E]_0 \cdot [S]_0}{[S]_0 + K_M};$$

$$v_0 = 75 \cdot 2.5 \cdot 10^{-6} \cdot 8 \cdot 10^{-4} / (8 \cdot 10^{-4} + 24 \cdot 10^{-6}) = 1.5 \cdot 10^{-7} / 8.24 \cdot 10^{-4} = 1.8 \cdot 10^{-4} \text{ M/s}.$$

Answer: $1.8 \cdot 10^{-4} \text{ M/s}$.

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