What is the pOH if the Molarity of NaOH is $2.3 \cdot 10^{-11}$.

Answer:

NaOH is a strong base. Therefore, its pOH is correlated with its molarity directly:

pOH = - lg [OH$^{/}$] = -lg [$2.3 \cdot 10^{-11}$] = 10.64

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