

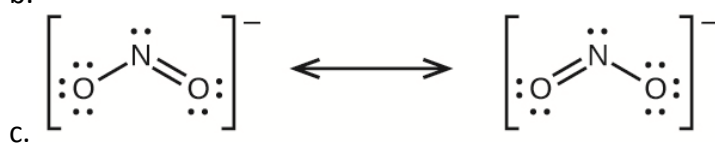
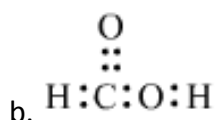
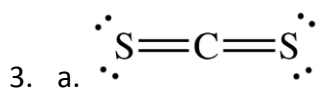
## Answer on Question #73572 – Chemistry – General Chemistry

1. Use Lewis dot symbols to represent the formation of barium hydride.
2. Which of the following bond is covalent, polar covalent, and ionic?
  - a. the bond in CsCl
  - b. the bond in H<sub>2</sub>S
  - c. the NN bond in H<sub>2</sub>NNH<sub>2</sub>
3. Write the lewis structure for
  - a. Carbon Disulfide (CS<sub>2</sub>)
  - b. Formic acid (HCOOH)
  - c. Nitrite ion (NO<sub>2</sub><sup>-</sup>)

### Solution:



- 1.
2.
  - a. ionic
  - b. polar covalent
  - c. covalent



Answer provided by <https://www.AssignmentExpert.com>