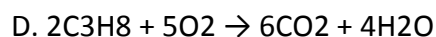
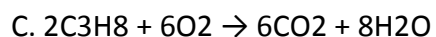
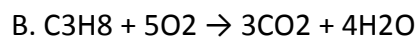
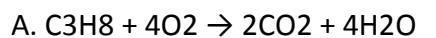


Question #72655, Chemistry / General Chemistry / Completed

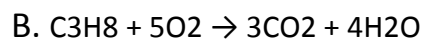
A gas grill burns Propane (C_3H_8) in the presence of more than sufficient Oxygen (O_2). This reaction produces water vapour and Carbon Dioxide. The temperature and pressure conditions are such that 1 mole of each gas occupies 1 litre of volume.

What is the balanced equation for this reaction?



E. None of the Above

Answer



	Before reaction:	After reaction:
C	3	3
H	8	8
O	10	10

Answer provided by AssignmentExpert.com