Question \#72324, Chemistry / Inorganic Chemistry
A metal x forms two different chloride each 12.70 g of chloride A and 16.30 g of chloride B contain 7.10 g and 10.70 g of chlorine respectively. Which gravimetric law would it be in accordance with

## Solution:

It is in accordance with the law of multiple proportions. It says that If two elements form more than one compound between them, then the ratios of the masses of the second element which combine with a fixed mass of the first element will be ratios of small whole numbers

|  | A | B |
| :---: | :---: | :---: |
| Total mass | 12,7 | 16,3 |
| Mass of Cl | 7,1 | 10,7 |
| Ratio of Cl | 0,559 | 0,656 |
| Ratio of Me | 0,441 | 0,344 |

A:
Ratio of Me : Ratio of $\mathrm{Cl}=0,441: 0,559=1: 1,267$
B:
Ratio of Me : Ratio of $\mathrm{Cl}=0,344: 0,656=1: 1,907$
$1,276: 1,907=1: 1,5=2: 3$

## Answer:

The law of multiple proportions

