## \#69224 Chemistry, Other

$3 \times 10^{8}$. Photoshop a certain light radiation are found to produce 1.5 j . Calculate wavelength.

Answer:
$\mathrm{E} \boldsymbol{\lambda}=\mathrm{hc}$
$\mathrm{h}=6.626 \times 10^{-34} \mathrm{~J} \cdot \mathrm{~s}$
$1.5 \cdot \lambda=6.626 \times 10^{-34} \cdot 3.00 \times 10^{8}$
$\lambda=\left(6.626 \times 10^{-34} \cdot 3.00 \times 10^{8}\right) / 1.5=19.88 \cdot 10^{-26} \mathrm{~m}$
Answer provided by https://www.AssignmentExpert.com

