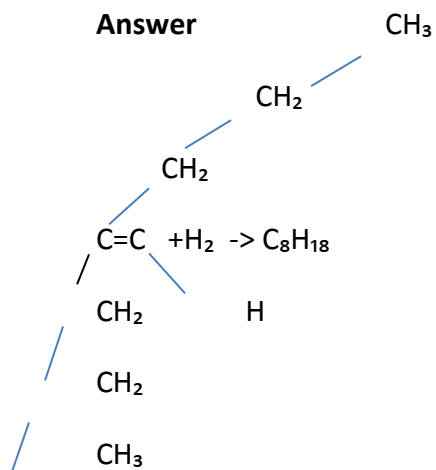


Answer on Question #65313, Chemistry / General Chemistry

The figure shows a scheme of the hydrogenation reaction. The reactant structure includes two double-bonded carbons, with a $-\text{CH}_2\text{CH}_2\text{CH}_3$ substituent and an $-\text{H}$ atom attached to each of them. The substituents are trans to each other. This hydrogenation is performed by using H_2 and a platinum catalyst. What is the product?

Answer



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