Question #64784, Chemistry / Inorganic Chemistry

Starting from Lewis structure, determine the hybridization types of the central atom of $TeCl_4$ and ICl_4

Answer:

```
TeCl<sub>4</sub>
:Cl:
:Cl: Te: Cl:
:Cl:
```

6 electrons on Te. The hybridization type is sp³d

ICl₄-

```
:C1:
:C1:
:C1:
:C1:
```

8 electrons on I. The hybridization type is $\mbox{sp}^{3}\mbox{d}^{2}$

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