

## Answer on Question #64378 - Chemistry - General Chemistry

### Task:

If  $Q = -91$  kJ for a certain, that process is called.

### Solution:

Since  $Q < 0$ , then the process is accompanied by absorption of heat. It is typical for an ENDOTHERMIC PROCESS.

For endothermic reactions, the enthalpy change of the internal energy and have positive values ( $\Delta H > 0$ ,  $\Delta U > 0$ ), thus the reaction products contain more energy than the original components.

Answer provided by [AssignmentExpert.com](http://AssignmentExpert.com)