

## Answer on the question #59879, Chemistry / Other

### Question:

Calculate the weight/volume percent composition, or %(W/V), of 455 ml of solution containing 15.0 g of sucrose.

### Solution:

Weight/volume percent is defined as:

$$w/v\% = \frac{\text{mass of the solute (g)}}{\text{volume of solution (mL)}} \cdot 100\%$$

$$w/v\% = \frac{15.0 \text{ g}}{455 \text{ mL}} \cdot 100\% = 3.3 \%$$

**Answer:** 3.3%