

Answer on Question #53108 – Chemistry – General chemistry

Task: what is the rate of reaction?

Answer:

The rate of reaction is the speed of decreasing/increasing of reactant/product concentration during the reaction. It is the measure of reaction intensiveness or, in other words, alterations of reactant and products concentrations in a definite time.

There is the formula for the rate of reaction:

$$v_R = \frac{\Delta c_P}{\Delta t}$$

where v_R is the rate of reaction,

Δc_P is the Alteration of product concentration;

Δt – is time.