## Answer on Question \#52374 - Chemistry - Inorganic Chemistry

## Question

What is the concentration, in $\mathrm{g} / \mathrm{mL}$, of a solution made by dissolving 75.0 g of phosphoric acid in 50.0 mL of solution?

## Answer:

The concentration of the solution is:

$$
C=\frac{m}{V}=\frac{75.0}{50.0}=1.5 \mathrm{~g} / \mathrm{ml}
$$

Answer: $1.5 \mathrm{~g} / \mathrm{ml}$

