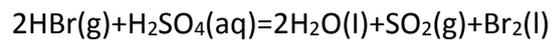


Answer to Question #51582, Chemistry, Organic Chemistry



From this equation work out the oxidation states of bromine and sulphur before and after the reaction. from this which has been oxidised and which reduced.

Answer:

	before the reaction	after the reaction
Br oxidised	HBr, Br ⁻¹	Br ₂ , Br ⁰
S reduced	H ₂ SO ₄ , S ⁺⁶	SO ₂ , S ⁺⁴

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