## Answer on the question \#50371, Chemistry, Other

## Question:

What is the pH of a solution whose $[\mathrm{H} 3 \mathrm{O}+]$ concentration is $3.2 \times 10-4 \mathrm{M}$ ?

## Answer:

The pH of the solution is:

$$
p H=-\lg [H 3 O+]=3.5
$$

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