Answer on Question #50204 - Chemistry – Other

Question

determining molar mass $Pb(NO_3)_2$

Answer:

Molar mass of $Pb(NO_3)_2$ equals:

$$M(Pb(NO_3)_2) = M(Pb) + 2(M(N) + 3M(O)) = 207.2 + 2(14.0 + 3 \cdot 16.0) = 331.2 \; g/mole$$

Answer: 331.2 g/mole