

Answer on Question #50203, Chemistry, Other

Determining the molar mass CaI_2

Solution:

$$M_r = \sum n_i \times A_{ri}$$

$$M_r(\text{CaI}_2) = A_r(\text{Ca}) + 2 \times A_r(\text{I}) = 40.078 + 2 \times 126.90447 = 293.88694$$

Answer:

293.88694

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