## Answer on the question \#49038, Chemistry, Physical Chemistry

## Question:

The two electrons having kinetic energies 16 eV and 49 eV respectively. What is the ratio of their wave lengths ?

## Solution:

The relation between energy and wavelength of the electron is:

$$
\lambda=\frac{\mathrm{h}}{\sqrt{2 \mathrm{mE}}}
$$

Then, the ratio of electron wavelength is:

$$
\frac{\lambda_{1}}{\lambda_{2}}=\frac{\mathrm{h}}{\sqrt{2 \mathrm{mE}_{1}}} * \frac{\sqrt{2 \mathrm{mE}_{2}}}{\mathrm{~h}}=\frac{\sqrt{\mathrm{E}_{2}}}{\sqrt{\mathrm{E}_{1}}}=\frac{\sqrt{49}}{\sqrt{16}}=\frac{7}{4}=1.75
$$

Answer: 1.75

