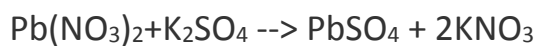


Question #47972, Chemistry, Physical Chemistry

A double-replacement reaction takes place when aqueous K_2SO_4 reacts with aqueous $Pb(NO_3)_2$. You would expect one of the products of this reaction to be:

- A. K_2S .
- B. CNO_3 .
- C. $NaPb$.
- D. $PbSO_4$.

Answer:



D. $PbSO_4$