Question \#47932, Chemistry, Inorganic Chemistry An HCl solution has a $[\mathrm{H}+$ ]of $4.4 \times 10-5 \mathrm{M}$. What is its pH ?

$$
p H=-\lg \left[H^{+}\right]=-\lg 4.4 * 10^{-5}=4.357
$$

