

Answer on Question #43792 - Chemistry - Other

Question:

The number of C - H bonds in methylbutane is?

Solution:

The structure formula of butane is $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CH}_3$. The methyl is CH_3 . We can add this part to create methylbutane in any places of butane. The molecular formula of methylbutane is C_5H_{12} . The number of C - H bonds in methylbutane equals number of hydrogen atoms. So, the number of C - H bonds in methylbutane are 12.

Answer: 12 C - H bonds.