## Question \#34828

What mass of water contains 42.0 g of oxygen?
Solution:
The mass portion of oxygen in water is $\frac{16.00}{1.01 \cdot 2+16.00}=0.888$, so mass of water which contains 42.0 g of oxygen is $42.0 / 0.888=47.3$

Answer: 47.3 g of water contains 42.0 g of oxygen.

