NaC or Na2C2 is sodium carbide and it's definitely ionic compound.

Several carbides are assumed to be salts of the **acetylide anion C_2^{2-}** (also called percarbide), which has a triple bond between the two carbon atoms. Alkali metals, alkaline earth metals, and lanthanoid metals form acetylides, e.g., sodium carbide **Na_2C_2**, calcium carbide CaC_2, and LaC_2. Lanthanoids also form carbides (sesquicarbides, see below) with formula M_2C_3. Metals from group 11 also tend to form acetylides, such as copper (I) acetylide and silver acetylide. Carbides of the actinide elements, which have stoichiometry MC_2 and M_2C_3, are also described as salt-like derivatives of C_2^{2-}.

The C-C triple bond length ranges from 109.2 pm in CaC_2 (similar to ethyne), to 130.3 pm in LaC_2 and 134 pm in UC_2. The bonding in LaC_2has been described in terms of La^{III} with the extra electron delocalised into the antibonding orbital on C_2^{2-}, explaining the metallic conduction.

http://en.wikipedia.org/wiki/Carbide