Reaction for this process is next:

$$C_2H_5OH + 3O_2 = 2 CO_2 + 3H_2O$$

As can you see the molar ratio is  $1:3 (C_2H_5OH : O_2)$ 

So if you have 15 mol of ethanol, you need 15\*3=45.0 mol of  $O_2$  to completely burn all your alcohol.