## Task:

Is glycerin mix with s.hydroxide solution safe? and is glycerin is heated then add s.hydroxide solution safe?

## Answer:

Glycerin with NaOH forms sodium monoglyceride or diglyceride, which is safe.

The chemical equation for this reaction is

 $\mathsf{CH}_2\mathsf{OH}-\mathsf{CHOH}-\mathsf{CH}_2\mathsf{OH}+\mathsf{NaOH}\Leftrightarrow\mathsf{CH}_2\mathsf{OH}-\mathsf{CHOH}-\mathsf{CH}_2\mathsf{ONa}+\mathsf{H}_2\mathsf{O}$ 

Or

 $\mathsf{CH}_2\mathsf{OH}-\mathsf{CHOH}-\mathsf{CH}_2\mathsf{OH}+\mathsf{2NaOH}\Leftrightarrow\mathsf{CH}_2\mathsf{OH}-\mathsf{CHONa}-\mathsf{CH}_2\mathsf{ONa}+\mathsf{2H}_2\mathsf{O}.$ 

When glycerin is heated the following reaction occurs

 $CH_2OH-CHOH-CH_2OH \Leftrightarrow CH_2 = CH-CHO + 2H_2O$  (heating to 280 °C)

It's dehydration reaction. The product of this reaction is acrolein. It is a colourless liquid with a piercing, disagreeable, acrid smell. The smell of burnt fat (as when cooking oil is heated to its smoke point).

Acrolein is toxic and is a strong irritant for the skin, eyes, and nasal passages.

Adding NaOH doesn't change anything.