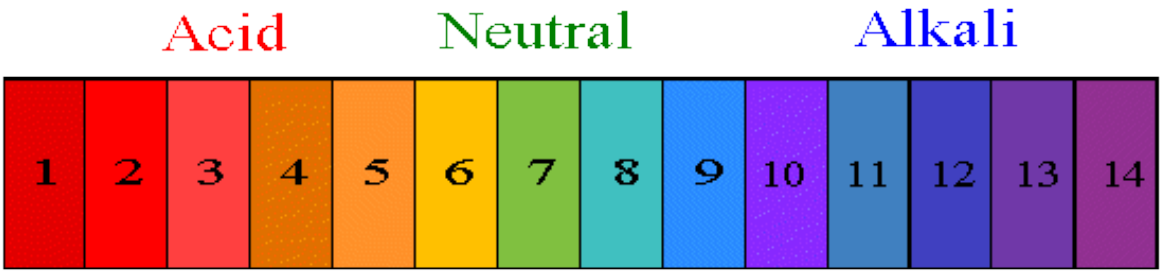
pH of 9,5 is really appropriate to alkali solution.



As you can see pH of 7 is neutral, but Dr.Sanchez have alkali water in her frog pond. Alkali means that there are some extra OH- ions in water, and only one way to neutralize these ions is adding some H+ ions and it is adding of acid. Why?

When solution is neutral it means that concentration of H+ and OH- is in equilibrium, as we said alkali means that there are some extra OH- ions. The reaction of neutralization is:

$$H^+ + OH^- \Leftrightarrow H_2O$$

So any acidic compound is acceptable for decreasing pH if only it is not poison for frogs of course.