It is better to use table for answering this question:

№	Volume in micro-litres	Volume in litres	Amount in moles	Concentration Mol/L (M)
1	500	0,0005	0,00025	0,025
2	1000	0,001	0,0005	0,05
3	1500	0,0015	0,00075	0,075
4	2000	0,002	0,001	0, 1

- 1 micro-litres = 1.0×10^{-6} l
- Amount is given concentration multiplied by volume in litres:

n=CV, C is 0.5M

- New concentration is amount multiplied by 1000 ml (one litre) / 10 ml