

$\mu\text{g/L}$  is micrograms per liter of any compound, it's a measure of the concentration in general (1 microgram per liter =  $1.0 \times 10^{-6}$  gram per liter ,  $1 \mu\text{g/l} = 1.0 \times 10^{-6} \text{ g/l}$ )

$\mu\text{g P L}^{-1}$  is total phosphorus concentration in  $\mu\text{g/L}$  .

That's all the difference.