It is possible to find concentration of acid from pH value.

$pH = -\log[H^+]$

$[H^+] = 10^{-pH}$

$[H^+] = 10^{-3} = 0.001$ M

After dilution concentration is $0.001/100 = 10^{-5}$

The new pH is :

$[H^+] = 10^{-5} = 10^{-pH}$

$pH = 5$