Molecular weight of water (H_2O) is: Mw= 18.015 g/mole

Amount of substance is : n = m/Mw.

n = 652g / 18.015 g/mol =36,19 mole

In chemistry and physics, the Avogadro constant (symbols: NA) is defined as the number of constituent particles (usually atoms or molecules) in one mole of a given substance. It has dimensions of reciprocal mol and its value is equal to 6.022×10^{23} mol^{-1}.

If 1 mole includes 6.022×10^{23} molecules, then 36,19 mole includes:

N = 36,19 * 6.022×10^{23} =$2,18 \times 10^{25}$ molecules