How can i create dilute sulphuric acid with sulpher powder.

Solution:

The process of receipt of sulphuric acid from sulphur includes three stages:

1) incineration of sulphur at surplus of air :

$$S + O_2 = SO_2$$

2) contact the oxidation of SO₂ to SO₃:

$$S + O_2 = SO_2$$

A process passes with by a catalyst on the basis of vanadium oxide V_2O_5 and at a temperature $420\text{-}550^{0}\text{C}$.

3) absorption of SO₃:

$$n SO_3 + H_2O = H_2SO_4 + (n-1)SO_3+Q$$

if n <1, we get diluted sulfuric acid.