

DIFFUSION	OSMOSIS
Diffusion is the movement of particles from an area of higher concentration to lower concentration.	Osmosis is the movement of solvent particles across a semipermeable membrane from a dilute solution into a concentrated solution.
Diffusion can occur in any medium, whether it is liquid, solid, or gas.	Osmosis occurs only in a liquid medium.
Hydrostatic pressure and turgor pressure do not normally apply to diffusion.	Hydrostatic pressure and turgor pressure oppose osmosis.
The movement in diffusion is to equalize concentration throughout the system.	The movement in osmosis seeks to equalize solvent concentration